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Chapter 15 Acids ...

Chapter 15: Acids and Bases
Acids and Bases
Arrhenius Definitions:
acids - compounds that produce an increase in $[H^+]$ when dissolved in water
bases - compounds that produce an increase in $[OH^-]$ when dissolved in water
Lewis Definitions:
acids - electron pair acceptors
bases - electron pair donors
Brønsted-Lowry

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Definitions: acids - H+ donors

Chapter 15: Acids and Bases Acids and Bases

Chapter 15 Acids and Bases. strong acid. strong base. weak acid. weak base. an acid that ionizes completely in solvent. a base that ionizes completely in a solvent. an acid that releases few hydrogen ions in aqueous solution. a base that

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releases few hydroxide ions in aqueous solution.

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454 CHAPTER 15

Formula Acid name HF

hydrofluoric acid HCl

hydrochloric acid HBr

hydrobromic acid HI

hydriodic acid H₂S

hydrosulfuric acid

TABLE 15-1 Names of

Binary Acids FIGURE

15-2 A strip of pH

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paper dipped into vinegar turns red, showing that vinegar is an acid.

CHAPTER 15 Acids and Bases

Chapter 15 Aqueous Equilibria: Acids and Bases. Instructor's Resource Materials (Download only) for Chemistry, ...

Conjugate Acid-Base Pairs: Chemical species whose formulas differ only by one hydrogen

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ion, H^+ Brønsted-Lowry Acid: A substance that can transfer hydrogen ions, H^+ . In other words, a proton donor

Chapter 15 Equilibria: Acids and Bases

Chapter 15 - Acids, Bases, & Salts. Lewis Acid. Lewis Base. Bronsted Lowry Acid. Bronsted Lowry Base. Anything that accepts a pair of electrons.

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Anything that donates a pair of electrons in a reaction. Anything that donates a proton, H^+ , in a reaction. Anything that accepts a proton in a reaction.

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Chapter 15 Acid and
Bases Vocabulary.
STUDY. Flashcards.
Learn. Write. Spell.
Test. PLAY. Match.

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Gravity. Created by.
sumbeautee.

Chemistry vocabulary
on acids and bases.

Terms in this set (21)
acid. a substance that
produced hydrogen
ions in aqueous
solutions. base. a
substance that
produced hydroxide
ions in aqueous
solutions.

**Chapter 15 Acid and
Bases Vocabulary
Flashcards | Quizlet**

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Chemistry Chapter 15
Acids and Bases Book
Notes. Sec 15.1:
Heartburn: Sec 15.2:
The Nature of Acids
and Bases: Properties
of Acids-sour taste,
ability to dissolve many
metals, ability to turn
blue litmus paper red,
and ability to neutralize
bases

**Chemistry Chapter
15 Acids and Bases
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CHAPTER 14 REVIEW

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Acids and Bases

SECTION 1 SHORT

ANSWER Answer the following questions in the space provided. 1.

Name the following compounds as acids:
sulfuric acid a. H_2SO_4
sulfurous acid b. H_2SO_3
hydrosulfuric acid c. H_2S
perchloric acid d. HClO_4
hydrocyanic acid e. hydrogen cyanide 2. H

14 Acids and Bases

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Chapter 15 Acids and
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Questions 1) _____ is

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found in carbonated beverages due to the reaction of carbon dioxide with water.

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Acids Bases Chapter
15 Review: Acids and
Bases Section 151
Bronsted Acids and
Bases Acids -
molecules that can lose
H⁺ (proton donors)

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making H_3O^+ in water
(acids lose H^+) Bases -
molecules than can
gain H^+ (proton
acceptors) and/or
make OH^- in water
(bases)

[PDF] Chapter 15 Acids Bases Section 2 Answers

Acids and Bases:

Chapter 14 & 15 . HW:

•Read Ch 14: •Fill in as
much of the acid base
table as you can, as
you read . Acid base

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conductivity and reactivity Conductivity, Reac'tivity,, ... 3. Work on ch 14-15 review (due Wednesday, test Friday!) • Solutions in which [H

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answers to Chapter 14
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Learning

Chapter 14 - Acids and Bases - Review Questions - Page 699: 2

Aside from a mention
of a textbook page,
this chemistry

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assignment is a suitable review of acids and bases. It begins by addressing conjugate pairs and acid base reactions.

Neutralization and amphoteric properties are also dealt with. The worksheet concludes with two real-life word problems.

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